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e4 Mathematics for Physical Chemistry the sine and cosine to the appropriate number of digits. $(31) 2^\circ \text{ rad}$
 $360 = 0.54 \text{ rad}$ $\sin(30.5^\circ) = 0.5075$ $\sin(31.5^\circ) = 0.5225$ $\sin(31^\circ) = 0.51$ $\cos(30.5^\circ) = 0.86163$ $\cos(31.5^\circ) = 0.85264$ $\cos(31^\circ) = 0.86$
15. Some elementary chemistry textbooks give the value of R, the ideal gas constant, as 0.0821 l atm K⁻¹ mol⁻¹. a.

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Solutions to exercises 1B.1(b) The mean speed is $\sqrt{8RT/v_{\text{mean}} M}$ The mean translational kinetic energy

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is $m \sqrt{3RT/2M}$ $\sqrt{3kT/2M}$ The ratios of species 1 to species 2 at the same temperature are $E_{k,1}/E_{k,2}$, $v_{mean,1}/v_{mean,2}$ (i) $1/2$ $m v_1^2/M_1$ $v_{mean,H}^2/2$ and $2 v_{mean,Hg}^2/200.6$ 4.003 $1/2$ $2 m v_{rms}^2/E_k$ $1/E_k$ $2/7.079$
(ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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