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Answer the following questions in the space provided. 1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ a. What is the value of the coefficient x in this equation? 40.07 g/mol

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b. What is the molar mass of C_3H_4 ? 2
mol O 2:1 mol H 2O c. What is the mole
ratio ...

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Stoichiometry Answers Mix & match

(both simple stoichiometry and limiting

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reagent problems) Units to use (select at least one): Grams Moles Particles (e.g. atoms/molecules/formula units) Chemical formulas or names: Formulas only Names only Mix & match formulas &

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Answers
Stoichiometry CHAPTER 9 REVIEW

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Write the answer on the line to the left.

Show all your work in the space provided.

1. _____ The actual yield of a reaction is 22 g and the theoretical yield is 25 g.

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Calculate the percentage yield. 2. 6.0 mol

Answers

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Stoichiometry. composition stoichiometry deals with the mass relationships of elements in compounds. reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

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HOMEWORK 9-2 (pp. 280-282)

VOCABULARY Complete each sentence.

1. ... the provided a 1 MODERN
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