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Unit 6 - Test (Chemical Compound) In chemistry, a compound refers to an entity that consists of two or more atoms, with at least two of those atoms from different chemical elements. These atoms associate via chemical bonds.

### Unit 6 - Test (Chemical Compound) - ProProfs Quiz

The lines/curves indicate the maximum amount of solute which can be dissolved in 100g of water at each temperature. We call this a saturated solution.. Any amount above the curve is said to be

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supersaturated; there is more solute than the solvent can hold, therefore the excess precipitates out/settles to the bottom.. Any amount below the line is called unsaturated, meaning more solute can be ...

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UNIT 6: RADIOACTIVITY AND NUCLEAR CHEMISTRY 5. A skeleton of a mammoth was analysed and found to contain 21.2% of the carbon-14 found in living bone. The half-life of carbon-14 is 5730 years. What is the most likely year of the mammoth's death? Use  $\log(N/N_0) = -t \log_2(1/2)$   $N/N_0 = 0.212$ ,  $t_{1/2} = 5730$  years; hence find  $t$  This year -  $t$  = year of death

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UNIT 6 - REDOX REACTIONS 6 • The oxidation number of an atom is the charge that would exist on an individual atom if the bonding were completely ionic • In simple ions, the oxidation number of the atom is the charge on the ion: - Na<sup>+</sup>, K<sup>+</sup>, H<sup>+</sup> all have an oxidation number of +1 - Mg<sup>2+</sup>, Ca<sup>2+</sup>, Pb<sup>2+</sup> all have an oxidation number of +2 - Cl<sup>-</sup>, Br<sup>-</sup>, I<sup>-</sup> all have an oxidation number of -1

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## GCSE Chemistry (Single Science) - BBC Bitesize

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